Tetrahedron Letters 50 (2009) 3296–3298

Contents lists available at ScienceDirect

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet

steps from known  $\beta$ -lactam (8) derived from  $D$ -mannitol triacetonide.

# Enantioselective total synthesis of (2S,3R,4R)-D-xylo-phytosphingosine from substituted azetidin-2-one

Ganesh Pandey \*, Dharmendra Kumar Tiwari

Division of Organic Chemistry, National Chemical Laboratory, Dr. Homi Bhabha Road, Pune 411 008, India

## article info

## **ABSTRACT**

Article history: Received 10 January 2009 Revised 2 February 2009 Accepted 9 February 2009 Available online 12 February 2009

Keywords: Enantioselective synthesis Sphingosine b-Lactam Wittig reaction

Sphingolipids which play crucial roles in many physiological  $processes<sup>1</sup>$  $processes<sup>1</sup>$  $processes<sup>1</sup>$  comprise sphingoid bases possessing long-chain aliphatic 2-amino-1,3-diol backbones. Sphingolipids such as Ceramide (1), sphingomyelins (2), and glycosphingolipids (3) are important membrane components of all the eukaryotic cells, plasma membranes, and intramolecular organelles playing vital roles in a number of cellular events including cell growth, differentiation, adhesion, and neuronal repair. They have also been proved to play a prominent role in cell signaling (Fig.  $1$ ).<sup>[2](#page-1-0)</sup>

Phytosphingosines, characterized by the 2-amino-1,3,4 triol head group, are the most important naturally occurring sphingolipids prevalent in microorganisms, plants, and many mammalian tissues such as brain, hair, intestines,<sup>3</sup> uterus,<sup>4</sup> liver,<sup>5</sup> skin,<sup>[6](#page-1-0)</sup> and blood plasma.[7](#page-1-0) They are also found in human kidney cerebrosides and some cancer cell types. $8$  In addition to being base components of sphingolipids in membranes, phytosphingosines themselves are found to be bioactive lipids. For example, phytosphingosine (5) is a potential heat stress signal in yeast cells<sup>[9](#page-1-0)</sup> and D-erythro-sphingosine  $(6)$  shows promising protein kinase inhibitory activity.<sup>10</sup> It has also been established that various diastereomers of sphingosines exhibit different activities and metabolism.<sup>11</sup> This subtle variation in biological activities over a range of diastereomers has led to the synthesis of all the diastereomers of sphingosines. This fact is very well reflected in the spurt of publications dealing with the synthesis of sphingosines.<sup>8,12-14</sup>

b-Lactams are part structures of the most widely used antibiotics and remain unsurpassed in their contribution to medicinal science as invaluable life-saving drugs. In addition to their medicinal values, b-lactams have also proved to be invaluable precursors for the syntheses of various natural products, $15-17$  owing to the flexi-bility available for selective ring-cleavage reactions. Banik et al.<sup>[18](#page-2-0)</sup> have utilized this property of  $\beta$ -lactam for the synthesis of nonnatural (-)-polyoxamic acid. Furthermore, β-lactam skeletons have also been used in the synthesis of sphingosines through their ring-opening reaction using either long alkyl Grignard reagents<sup>14a</sup>,  $n$ -tetradecyl p-toluenesulfonate/n-BuLi<sup>19</sup> or phosphonate-stabilized carbanion $^{20}$  $^{20}$  $^{20}$  as nucleophiles.

Enantiomerically pure (2S,3R,4R)-D-xylo phytosphingosine is synthesized in 36% overall yield in seven



**(2S,3R,4R)-xylo -phyto-sphingosine (4) R = H, Ceramide (1) R = Phosphocholine, Sphingomyeline (2) R = Sugars, Glycosphingolipid (3)**







- 2009 Elsevier Ltd. All rights reserved.

Corresponding author. Tel.: +91 20 25902627; fax: +91 20 25902628. E-mail address: gp.pandey@ncl.res.in (G. Pandey).

<sup>0040-4039/\$ -</sup> see front matter © 2009 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2009.02.050

<span id="page-1-0"></span>

**Scheme 1.** Reagents and conditions: (a)  $PPh_3BrC1_3H_{27}$ , n-BuLi, dry THF, 0 °C, 1 h, 70%; (b) HCOONH4, Pd/C (10%), MeOH, reflux, 4 h; 80%; (c) LAH, dry THF, reflux, 4 h, 93%; (d) H<sub>2</sub>, Pd/C (10%), MeOH, (Boc)<sub>2</sub>O, rt 10 h, 93%; (e) NaIO<sub>4</sub>, EtOH: H<sub>2</sub>O (1:1) rt, 30 min, 89%; (f) NaBH<sub>4</sub>, dry. MeOH, 0 °C to rt, 10 h; 92%; (g) TFA/H<sub>2</sub>O (20:1), DCM, 0 °C, 3 h; (h) Ac<sub>2</sub>O, dry Py, DMAP (cat) 91%.

While all these reports present attractive approaches to sphingosines, the installation of the tetradecyl chain has not been easy and has often produced a mixture of different products.<sup>[21](#page-2-0)</sup> Therefore, it was felt that a route which introduces the long tetradecyl chain in sphingosines before the opening of the  $\beta$ -lactam ring could be an attractive strategy. Toward this end, we have devised a synthetic route for  $D-xy$ lo-phytosphingosine, starting from  $\beta$ -lactam 8 derived from p-mannitol triacetonide 7 and report herein our successful preliminary endeavor (Scheme 1).

The required  $\beta$ -lactam 8 was synthesized in 70% yield using the reported protocol from <sub>D</sub>-mannitol triacetonide **7**.<sup>[18](#page-2-0)</sup> Subjecting **8** to Wittig olefination with a 13 carbon ylide in the presence of n-BuLi at 0 °C produced 9 in 70% yield as a cis, trans mixture (55:45), confirmed by  $^1\mathrm{H}$  NMR spectra. The geometrical isomers ratio was of no relevance to the planned synthetic sequence as the double bond was to be reduced in the immediate next step. It was also expected that during the reduction of the olefinic double bond of 9, O-debenzylation would occur. Accordingly, 9 upon transfer hydrogenation using ammonium formate and Pd/C (10%) in methanol furnished 10 in 80% yield. Compound 10 upon reduction with lithium aluminum hydride in THF, under reflux condition produced 11 in 93% yield having a vicinal diol moiety. N-Debenzylation of 11 by catalytic hydrogenation (10%, Pd/C) at atmospheric pressure of hydrogen followed by N-Boc protection gave 12 in 93% over two steps. The oxidative cleavage of 12 using sodium periodate in ethanol/water (1:1) solvent at room temperature yielded 13 in 89% yield. The subsequent reduction of 13 with sodium borohydride gave compound 14 (92%). The treatment of 14 with TFA/water (20:1) in dichloromethane at  $0 °C$  led to the concomitant N-Boc as well as acetonide deprotection affording 4 quantitatively. Since it was difficult to purify 4 by column chromatography, this product was acetylated using acetic anhydride/ pyridine with a catalytic amount of DMAP to obtain 15. Compound 15 was easily purified by silica gel column chromatography in 30% ethylacetate/petroleum ether as eluent.

The spectral data and specific rotation of 15 were in excellent agreement with the reported values.<sup>[22](#page-2-0)</sup>

In conclusion, an enantioselective synthesis of (2S,3R,4R)-D $xy$ lo-phytosphingosine is achieved from  $\beta$ -lactam derived from  $D$ mannitol triacetonide.

#### Acknowledgments

The authors thank the Department of Science and Technology, New Delhi, for financial support and CSIR, New Delhi, for a research fellowship to D. K. Tiwari.

#### References and notes

- 1. Merrill, A. H., Jr.; Sandhoff, K. Sphingolipids: Metabolism and Cell Signaling. In Biochemistry of Lipids, Lipoprotein, and Membranes; Vance, D. E., Vance, J. E., Eds.; Elsevier: New York, 2002; pp 373–407.
- 2. (a) Scneiter, R. Bioassays 1999, 21, 1004; (b) Kobayashi, E.; Motoki, K.; Yamaguchi, Y.; Uchida, T.; Fukushima, H.; Koezuka, Y. Bioorg. Med. Chem. 1996, 32, 133.
- 3. Okabe, K.; Keeman, R. W.; Schmidt, G. Biochem. Biophys. Res. Commun. 1968, 31, 137.
- 4. Takamatsu, K.; Mikami, M.; Kikuchi, K.; Nozawa, S.; Iwamori, M. Biochim. Biophys. Acta 1992, 1165, 177.
- 5. Barenholz, Y.; Gatt, S. Biochim. Biophys. Res. Commun. 1967, 27, 319.
- 6. (a) Wertz, P. W.; Miethke, M. C.; Long, S. A.; Stauss, J. S.; Owning, D. T. J. Invest. Dertmatol. 1985, 84, 410; (b) Schmidt, R. R. In Liposome Dermatics; Braun-Falco, O., Corting, H. C., Maibach, H. I., Eds.; Springer: Berlin, 1992; pp 44–56.
- 7. Vance, D. E.; Sweeley, C. C. J. Lipid Res. 1967, 8, 621.
- 8. For recent reviews on sphingosines/ceramides, see: (a) Liao, J.; Tao, J.; Lin, G.; Liu, D. Tetrahedron 2005, 61, 4715; (b) Curfman, C.; Liotta, D. Methods Enzymol. 1999, 311, 391; (c) Jeong-Ju, Park; Ji, Hyung Lee; Qian, Li; Kristine, Diaz; Young-Tae, Chang; Sung-Kee, Chung Bioorg. Chem. 2008, 36, 220–228.
- 9. (a) Dickson, R. C.; Nagiec, E. E.; Skrzypek, M.; Tillman, P.; Wells, G. B.; Lester, R. L. J. Biol. Chem. 1997, 272, 30196; (b) Schneiter, R. Bioessays 1999, 21, 1004.
- 10. (a) Brodesser, S.; Sawatzki, P.; Kolter, T. Eur. J. Org. Chem. 2003, 2021; (b) Vankar, Y. D.; Schmidt, R. R. Chem. Soc. Rev 2000, 29, 201.
- 11. (a) Inokuchi, J.-i.; Usuki, S.; Jimbo, M. J. Biochem. (Tokyo) 1995, 117, 766; (b) Motoki, K.; Kobayashi, E.; Uchida, T.; Fukushima, H.; Koezuka, Y. Bioorg. Med. Chem. Lett. 1995, 5, 705; (c) Pushkareva, M.; Chao, R.; Beilawska, A.; Merrill, H.; Crane, H. M.; Lagu, B.; Liotta, D.; Hannun, Y. A. Biochemistry 1995, 34, 1885.
- 12. For studies on the synthesis of phytosphingosine and its derivatives, see: (a) Sugiyama, S.; Honda, M.; Komori, T. Liebigs Ann. Chem. 1990, 1069; (b) Dondoni, A.; Fantin, G.; Fogagnolo, M.; Pedrini, P. J. Org. Chem. 1990, 55, 439; (c) Wild, R.; Schmidt, R. R. Tetrahedron: Asymmetry 1994, 5, 2195; (d) Murakami, T.; Minamikawa, H.; Hato, M. Tetrahedron Lett. 1994, 35, 745; (e) Nakashima, H.; Hirata, N.; Iwamura, T.; Yamagiwa, Y.; Kamikawa, T. J. Chem. Soc., Perkin Trans. 1 1994, 2849; (f) Kobayashi, S.; Hayashi, T.; Kawasuji, T. Tetrahedron Lett. 1994, 35, 9573; (g) Matsumoto, K.; Ebatu, T.; Matsushita, H. Carbohydr. Res. 1995, 279, 93; (h) Wild, R.; Schmidt, R. R. Liebigs Ann. 1995, 755; (i) Li, Y. L.; Mao, X. H.; Wu, Y. L. J. Chem. Soc., Perkin Trans. 1 1995, 1559; (j) Kemp, S. J.; Bau, J.; Pedersen, S. F. J. Org. Chem. 1996, 61, 7162; (k) Lin, G. Q.; Shi, Z. C. Tetrahedron 1996, 52, 2187.
- 13. (a) Yoda, H.; Oguchi, T.; Takabe, K. Tetrahedron: Asymmetry 1996, 7, 2113; (b) Murakami, M.; Ito, H.; Ito, Y. Chem. Lett. 1996, 185; (c) Wee, A. G. H.; Tang, F. Tetrahedron Lett. 1996, 37, 677; (d) Shimizu, M.; Wakioka, I.; Fujisawa, T. Tetrahedron Lett. 1997, 38, 6027; (e) Imashiro, R.; Sakurai, O.; Yamashita, T.; Horikawa, H. Tetrahedron 1998, 54, 10657; (f) Takikawa, H.; Muto, S.; Mori, K. Tetrahedron 1998, 54, 3141; (g) Li, S.; Pang, J.; Wilson, W. K.; Schroepfer, J., Jr. Tetrahedron: Asymmetry 1999, 10, 1697; (h) Murakami, T.; Toguchi, K. Tetrahedron 1999, 55, 989; (i) Shirota, O.; Nakanishi, K.; Berova, N. Tetrahedron 1999, 55, 13643; (j) Fernandes, R. A.; Kumar, P. Synthesis 2003, 1, 129.
- <span id="page-2-0"></span>14. (a) Kale, A. S.; Sakle, P. S.; Gumaste, V. K.; Deshmukh, A. R. A. S. Synthesis 2007, 17, 2631; (b) Olofsson, B.; Somfai, P. J. Org. Chem. 2003, 68, 2514; (c) Raghvan, S.; Rajandar, A.; Yadav, J. S. Tetrahedron Asymmetry 2003, 14, 2093; (d) Chaudhari, V. D.; Ajish Kumar, K. S.; Dhavale, D. D. Org Lett 2005, 7, 5805; (e) Lee, J.-M.; Lim, H.-S.; Chung, S.-K. Tetrahedron: Asymmetry 2002, 13, 343; (f) Lombardo, M.; Capdevila, M. G.; Pasi, F.; Trombini, C. Org. Lett. 2006, 8, 3303.
- 15. (a) Thomas, R. C. In Recent Progress in the Chemical Synthesis of Antibiotics; Lukacs, G., Ohno, M., Eds.; Springer: Berlin, 1990; p 553; (b) Palomo, C. In Recent Progress in the Chemical Synthesis of Antibiotics; Lukacs, G., Ohno, M., Eds.; Springer: Berlin, 1990; p 565; (c) Nagahara, T.; Kametani, T. Heterocycles 1987, 25, 729; (d) Palomo, C.; Cabre, F.; Ontoria, J. M. Tetrahedron Lett. 1992, 33, 4819.
- 16. (a) Jayaraman, M.; Deshmukh, A. R. A. S.; Bhawal, B. M. J. Org. Chem 1994, 59, 932; (b) Jayaraman, M.; Nandi, M.; Sathe, K. M.; Deshmukh, A. R. A. S.; Bhawal, B. M. Tetrahedron: Asymmetry 1993, 4, 609; (c) Jayaraman, M. J.; Deshmukh, A. R. A. S.; Bhawal, B. M. Tetrahedron 1996, 52, 8989.
- 17. (a) Holton, R. A.; Liu, J. H. Bioorg. Med. Chem. Lett. 1993, 3, 2475; (b) Denis, J.-N.; Greene, A. E.; Guénard, D.; Guéritte-Voegelein, F.; Mangatal, L.; Potier, P. J. Am.

Chem. Soc. 1988, 110, 5917; (c) Georg, G. I.; Cheruvallath, Z. S.; Harriman, G. C. B.; Hepperle, M.; Park, H. Bioorg. Med. Chem. Lett. 1993, 3, 2467.

- 18. (a) Banik, B. K.; Manhas, M. S.; Bose, A. K. J. Org. Chem. 1993, 58, 307; (b) Banik, B. K.; Mathur, C.; Wagle, D. R.; Manhas, M. S.; Bose, A. K. Tetrahedron 2000, 56, 5603.
- 19. Nakamura, T.; Shiozaki, M. Tetrahedron Lett. 1999, 40, 9063.
- 20. Lee, K. H.; Kim, E. K.; Pak, C. S. Tetrahedron Lett. 2002, 43, 9641.
- 21. Palomo, C.; Aizpurus, J. M.; Garcia, J. M.; Iturburu, M.; Odriozola, J. M. J. Org. Chem. 1994, 59, 5184.
- 22. Nakada, A. J.; Hashemzadeh, M.; Regina, C.; Howell, R. A. Org. Lett. **2002**, 4.<br>1719. Spectral data of compound (**15**): Mp 44–46 °C; [ $\alpha_{\text{II}}^{27}$  6.5 (c 0.86, CHCl<sub>3</sub>), {lit <sup>18</sup>[ $\alpha_{\text{II}}^{27}$  6.3 (c 0.86, CHCl<sub>3</sub> 954 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  5.77 (d, J = 9.5 Hz, 1H), 5.14 (dd, J = 6.6 4.3 Hz, 1H), 5.03 (dd,  $J = 12.7$ , 6.8 Hz, 1H), 4.51 (m, 1H), 4.01 (m, 2H), 2.08 (s, 3H), 2.05 (s, 3H), 2.04 (s, 3H), 2.00 (s, 3H) 152–1.60 (m, 2H), 1.22 (bs, 24H), 0.86 (t, J = 6.4 Hz, 3H), 13C NMR (75 MHz, CDCl3): 170.6, 170.1, 169.8, 72.2, 71.9, 62.9, 47.9, 31.9, 305.5, 29.7, 29.6, 29.3, 29.2, 24.8, 23.2, 22.7, 20.9, 20.7, 20.6, 14.1. MS (m/z): 486 (M<sup>+</sup>+1).